

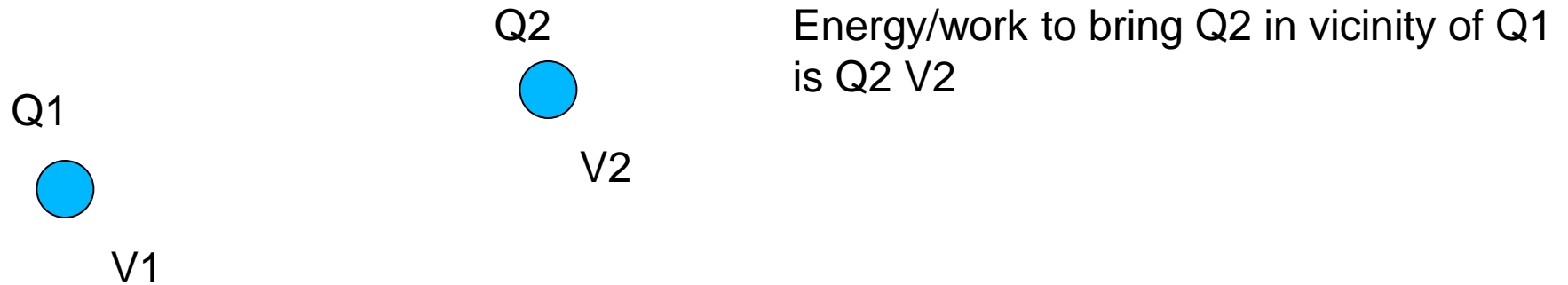
Topic 10: Physics of charged solutions: Applications
(chapter 9 in book)

Outline:

- What's the energetic cost of packing a protein that has charge?
- What's the energetic cost of packing charged DNA into a viral capsid?

Electrostatic energy:

How much energy is there to bring two charges together?

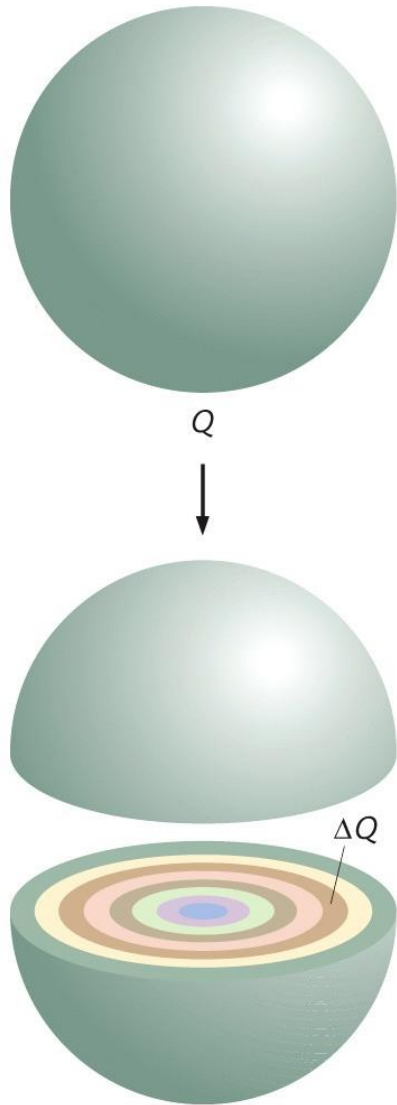


The total energy to bring these two charges together is: $\frac{1}{2}(Q1 V1 + Q2 V2)$

So in general:

$$U = \frac{1}{2} \sum Q_i V_i = \frac{1}{2} \int V(\mathbf{r}) \rho(\mathbf{r}) d^3\mathbf{r} = \int V(\mathbf{r}) dQ$$

Example: assembling a charged sphere



Let's use our formula to evaluate U for a sphere that has charge Q distributed uniformly over its volume.

The sphere gets assembled layer by layer. When we are about to add a layer at radius, r , with a charge, q enclosed inside the potential at this distance is:

$$V(r) = \frac{1}{4\pi D\epsilon_0} \frac{q}{r} = \frac{1}{4\pi D\epsilon_0} \frac{\rho \frac{4}{3}\pi r^3}{r}$$

So the energy to assemble is:

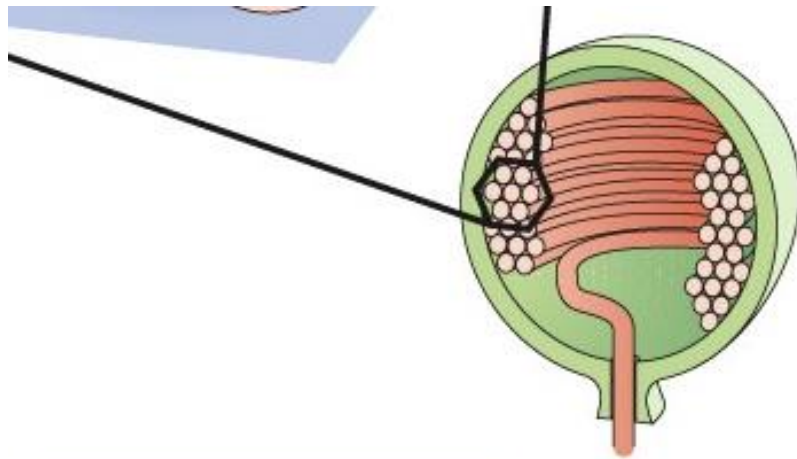
$$U = 4\pi \int \frac{1}{4\pi D\epsilon_0} \frac{\rho \left(\frac{4}{3}\right)\pi r^3}{r} \rho r^2 dr$$

so

$$U = \frac{1}{4\pi D\epsilon_0} \left(\frac{4}{3}\right) \pi 4\pi \rho^2 \frac{R^5}{5} = \frac{1}{4\pi D\epsilon_0} \frac{3}{5} \frac{Q^2}{R}$$

Figure 9.11 Physical Biology of the Cell, 2ed. (© Garland Science 2013)

First pass calculation: DNA in capsid



DNA in capsid

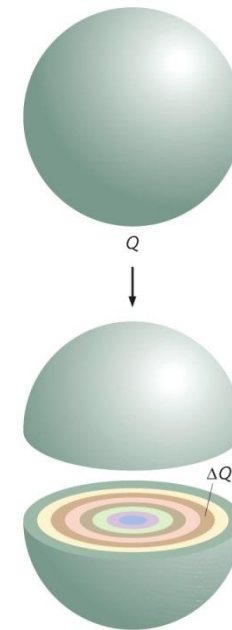
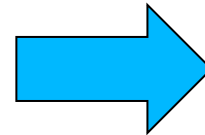


Figure 9.11 Physical Biology of the Cell, 2ed. (© Garland Science 2013)

If there is no screening, how much energy to put DNA into capsid? Assume charge Q of DNA is uniformly distributed in volume of capsid, so we can use our equation

For DNA in $\phi 29$ virus that has 20000 bp and there are 2 e / bp for DNA

$Q = 20000 * 2e = 400000 e$ and $R = 20 \text{ nm}$ using $D = 80$ for H₂O gives

so
$$U = \frac{1}{4\pi D \epsilon_0} \frac{3}{5} \frac{Q^2}{R} = 10^8 \text{ pN nm} \gg 200000 \text{ pN nm that was measured}$$

Resolution: screening due to counterions greatly reduces the potential energy

Energy to assemble a protein:

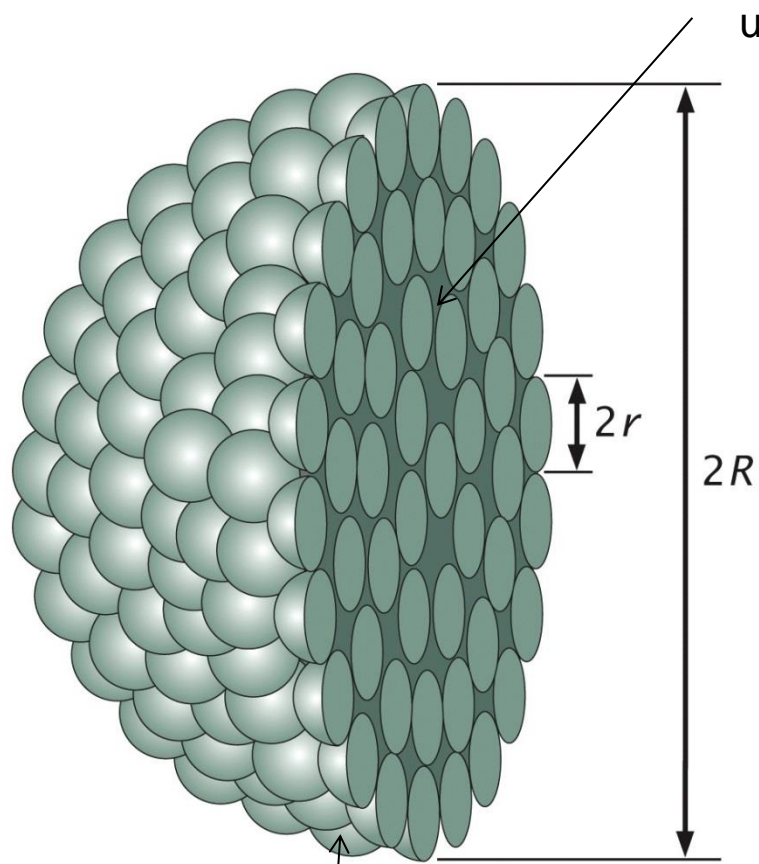


Figure 9.13 Physical Biology of the Cell, 2ed. (© Garland Science 2013)

How does the energy to assemble the protein depend on its size (i.e. the number of polar residues on its surface)?

$$U = \frac{1}{2} \int V(\mathbf{r}) \rho(\mathbf{r}) d^3\mathbf{r} = \frac{1}{2} \int \frac{1}{4\pi D \epsilon_0} \frac{Q_{tot}}{R} \frac{Q_{tot}}{4\pi R^2} dA$$

so

$$U = \frac{1}{2} \frac{1}{4\pi D \epsilon_0} \frac{Q^2}{R}$$

Now $Q = N e$ where $N = (4\pi R^2)/\pi r^2$

So the energy is

$$U = \frac{1}{2} \frac{e^2}{4\pi D \epsilon_0} \frac{R^3}{r^4}$$

Assembling proteins:

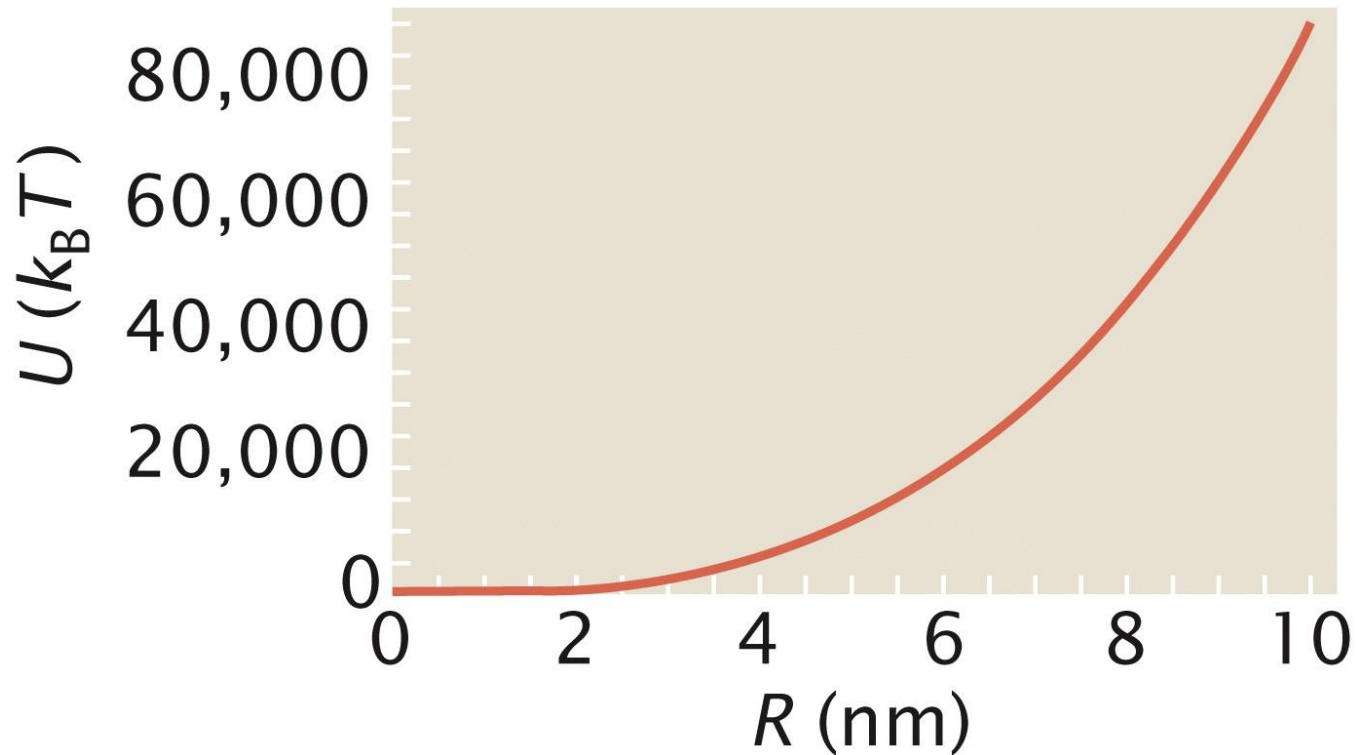
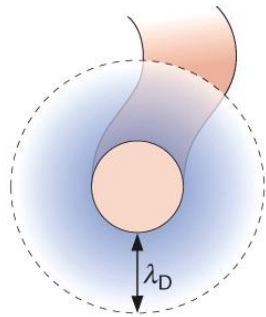


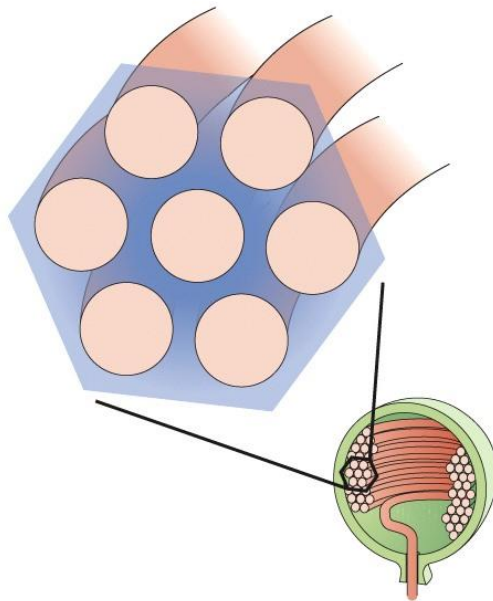
Figure 9.14 Physical Biology of the Cell, 2ed. (© Garland Science 2013)

Key is that for proteins of typical size ~ 2 nm, it does not cost much charging energy to assemble the object. Whereas larger objects ~ 10 nm and bigger are more costly.

Packing DNA into a virus: Redux



DNA in solution



DNA in capsid

We know that counterions screen the DNA

The Debye screening length is given by λ_D

So given this, what is the energy required to assemble into the capsid?

Now the system is effectively screened, and so the object is neutral.

The energy cost is the loss of entropy of the cations in going from the outside to the inside

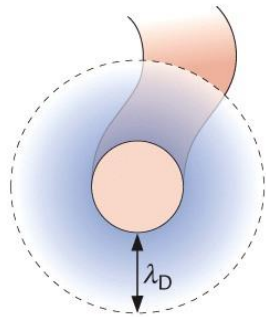
Recall our entropy for gas, $S \sim N \log(V)$

So the work done is $dG = (-kT N \log V_{cap}) - (-kT N \log V_{cloud})$

or

$$\Delta G = N k T \log \frac{V_{cloud}}{V_{cap}}$$

Packing DNA into a virus: Redux



DNA in solution

Since there were -40000 e for the DNA, there are $N = +40000$ e counterions

The volume of the cloud outside capsid is:

$$V_{cloud} = L \pi ((r + \lambda_D)^2 - r^2)$$

where $r = 2$ nm for DNA, and $\lambda \sim 1$ nm

The free volume for the counterions in the capsid is

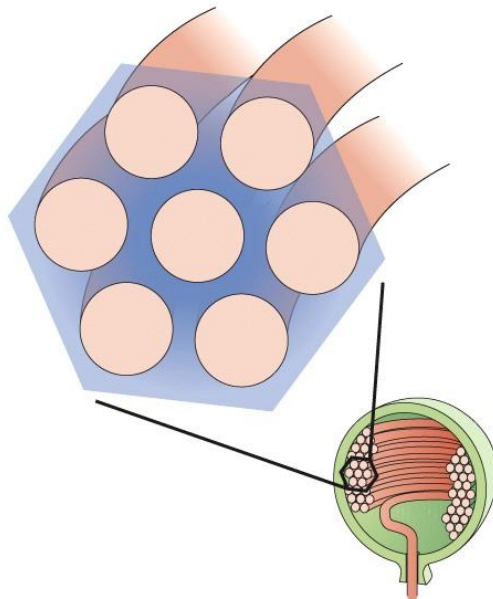
$$V_{capsid} = \frac{4}{3} \pi R^3 - L \pi r^2$$

where $R = 20$ nm for the phi29 virus.

Using the previous equation and these numbers

$\Delta G = 265000$ pN nm ~ 200000 pN nm that was

measured experimentally.



DNA in capsid

Summary:

- Looked at the energy required to assemble charge
- For DNA packed into capsid, without counterions a significant energy would be required to overcome electrostatic energy
- Assembling small charged objects like proteins is not that costly
- Counterions make DNA effectively neutral and the energy cost is mostly the entropic cost of moving the counterions into the capsid
- Compare that to the almost negligible entropic cost for the DNA polymer itself (you did that on A 5)