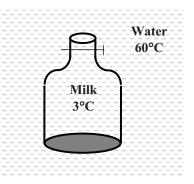
**11-18** A thin-walled glass containing milk is placed into a large pan filled with hot water to warm up the milk. The warming time of the milk is to be determined.

Assumptions 1 The glass container is cylindrical in shape with a radius of  $r_0 = 3$  cm. 2 The thermal properties of the milk are taken to be the same as those of water. 3 Thermal properties of the milk are constant at room temperature. 4 The heat transfer coefficient is constant and uniform over the entire surface. 5 The Biot number in this case is large (much larger than 0.1). However, the lumped system analysis is still applicable since the milk is stirred constantly, so that its temperature remains uniform at all times.

**Properties** The thermal conductivity, density, and specific heat of the milk at 20°C are k = 0.598 W/m.°C,  $\rho = 998$  kg/m<sup>3</sup>, and  $c_p = 4.182$  kJ/kg.°C (Table A-15).



Analysis The characteristic length and Biot number for the glass of milk are

$$L_{c} = \frac{\mathbf{V}}{A_{s}} = \frac{\pi r_{o}^{2} L}{2\pi r_{o} L + 2\pi r_{o}^{2}} = \frac{\pi (0.03 \text{ m})^{2} (0.07 \text{ m})}{2\pi (0.03 \text{ m}) (0.07 \text{ m}) + 2\pi (0.03 \text{ m})^{2}} = 0.01050 \text{ m}$$
$$Bi = \frac{hL_{c}}{k} = \frac{(240 \text{ W/m}^{2}.^{\circ}\text{C})(0.0105 \text{ m})}{(0.598 \text{ W/m}.^{\circ}\text{C})} = 4.21 > 0.1$$

For the reason explained above we can use the lumped system analysis to determine how long it will take for the milk to warm up to 38°C:

$$b = \frac{hA_s}{\rho c_p V} = \frac{h}{\rho c_p L_c} = \frac{240 \text{ W/m}^2.^\circ\text{C}}{(998 \text{ kg/m}^3)(4182 \text{ J/kg.}^\circ\text{C})(0.0105 \text{ m})} = 0.005477 \text{ s}^{-1}$$
$$\frac{T(t) - T_{\infty}}{T_i - T_{\infty}} = e^{-bt} \longrightarrow \frac{38 - 60}{3 - 60} = e^{-(0.005477 \text{ s}^{-1})t} \longrightarrow t = 174 \text{ s} = 2.9 \text{ min}$$

Therefore, it will take about 3 minutes to warm the milk from 3 to 38°C.

**11-19** A long copper rod is cooled to a specified temperature. The cooling time is to be determined. *Assumptions* **1** The thermal properties of the geometry are constant. **2** The heat transfer coefficient is constant and uniform over the entire surface.

*Properties* The properties of copper are k = 401 W/m·°C,  $\rho = 8933$  kg/m<sup>3</sup>, and  $c_p = 0.385$  kJ/kg·°C (Table A-24).

Analysis For cylinder, the characteristic length and the Biot number are

$$L_{c} = \frac{V}{A_{\text{surface}}} = \frac{(\pi D^{2}/4)L}{\pi DL} = \frac{D}{4} = \frac{0.02 \text{ m}}{4} = 0.005 \text{ m}$$
  

$$Bi = \frac{hL_{c}}{k} = \frac{(200 \text{ W/m}^{2}.^{\circ}\text{C})(0.005 \text{ m})}{(401 \text{ W/m}.^{\circ}\text{C})} = 0.0025 < 0.1$$
  

$$T_{i} = 100 \text{ °C}$$

Since Bi < 0.1, the lumped system analysis is applicable. Then the cooling time is determined from

$$b = \frac{hA}{\rho c_p V} = \frac{h}{\rho c_p L_c} = \frac{200 \text{ W/m}^2 \cdot ^\circ\text{C}}{(8933 \text{ kg/m}^3)(385 \text{ J/kg.}^\circ\text{C})(0.005 \text{ m})} = 0.01163 \text{ s}^{-1}$$

$$\frac{T(t) - T_{\infty}}{T_i - T_{\infty}} = e^{-bt} \longrightarrow \frac{25 - 20}{100 - 20} = e^{-(0.01163 \,\mathrm{s}^{-1})t} \longrightarrow t = 238 \,\mathrm{s} = 4.0 \,\mathrm{min}$$

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